## **Course Specifications**

Programme(s) on which the course is given: Chemistry

Major or Minor element of programmes: Minor Department offering the programme: Chemistry

**Department offering the course: Chemistry** 

**Prerequisite: CH278** 

Academic year / Level: Third

Date of specification approval: 2013

**A- Basic Information** 

Title: Practical analytical chemistry (3)

Code:

**CH3711** 

Credit Hours: 4 h Lecture:0.0

Tutorial: 4 Practicals: 2 Total: 8h

- **B- Professional Information** 
  - 1 Overall Aims of Course

Illustrate how we can determine the percent of metals in ore by dissolving and transfer it to precipitate accompanied by know chemical structure.

- 2 Intended Learning Outcomes of Course (ILOs)
  - a- Knowledge and Understanding:

The graduates should be able after completing the course to understand the basic steps of gravimetry.

- **b- Intellectual Skills** 
  - 1- Improve the formation of precipitate a given element in a solution
  - 2- Learn how to reduce coprecipitation, post precipitation and peptization errors
  - **3- Improve the means of separation of precipitate from its mother liquor**
- c- Professional and Practical Skills
  - 1- Setup of different experiments of gravimetry.
  - 2- Differentiate between drying and ignition methods

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3- Verdict of some laws of chemistry such as Ksp, Debye Huckel equation and chemical factor.

## d- General and Transferable Skills

- 1- Use some tools and apparatus such as gooch and ashless filter paper in filtration
- 2- Enhance the representation of scientific data
- 3- Reduce the experimental erro

## **3- Contents**

Topic	No. of hours	Lectu re	Tutori al/ Practic al
Determination of H <sub>2</sub> O in	8	_	8
BaCl <sub>2.</sub> n H <sub>2</sub> O			
<b>Determination of Ba as</b>	16	-	16
BaSO <sub>4</sub>			
<b>Determination of Mg as</b>		-	
MgNH <sub>4</sub> PO <sub>4</sub> .6H <sub>2</sub> 0	8		8
<b>Determination of Ni as the</b>		-	
glyoximate complex	16		16
<b>Determination of Pb as</b>	8	-	8
PbCrO <sub>4</sub>			
<b>Determination of Ca as</b>	8	-	8
CaC <sub>2</sub> O <sub>4</sub> .H <sub>2</sub> O			
<b>Determination of Al as Al<sub>2</sub>O<sub>3</sub></b>	16	-	16
<b>Determination of Fe as</b>	16	-	16
$Fe_2O_3$			

## 4- Teaching and Learning Methods

- 1 practical work
- 2-disscutions
- **5- Graduate Assessment Methods**

1 written examination to assess the understanding and comprehension

2- practical exam to assess the performance and profisionalism

**Assessment Schedule** 

Assessment 1 short exam (class activities) Week every two weeks

Assessment 2 mid-term (written and practical) Week 8

Assessment 3 final-term (written and practical) Week 13 and 14

**Weighting of Assessments** 

**Mid-Term Examination 20%** 

Final-term Examination 60%

Semester Work 20%

**Total** 100%

- **6- List of References** 
  - a- Essential Books (Text Books)
    - 1- Vogel
    - 2- Alexeyev
  - 7- Facilities Required for Teaching and Learning Experimental lab...

**Course Coordinator:** 

Head of Department: Prof. Dr. Adel Nassar

**Date:** / /